

Per- and Polyfluoroalkyl Substances in Cloud Water over the Northwestern Atlantic



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ACTIVATE STM
Tucson, AZ

13 – 15 November 2023

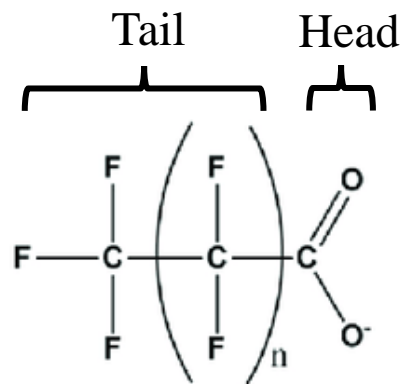
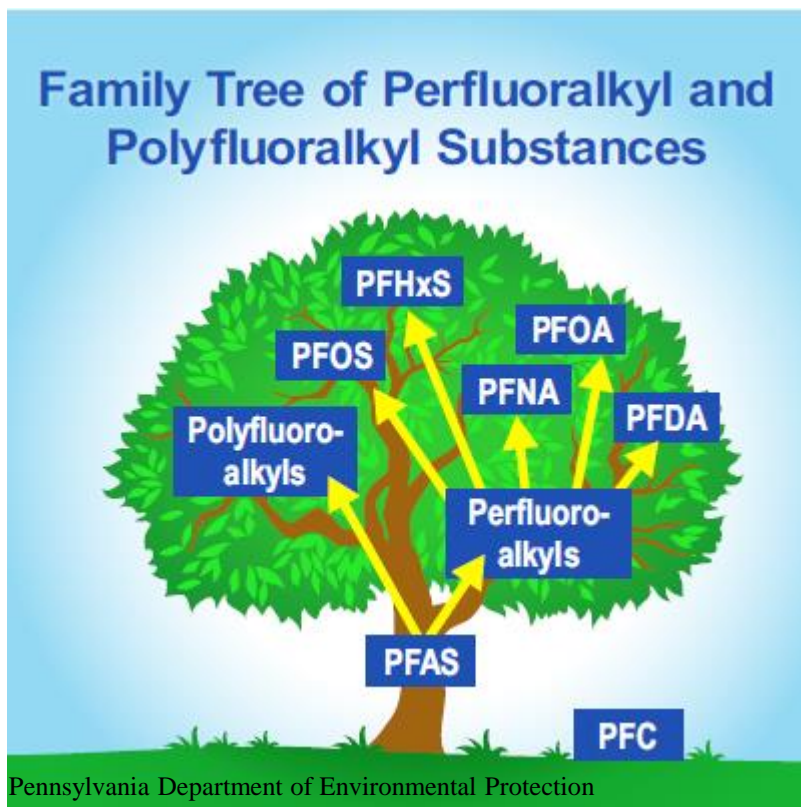


What are per- and polyfluoroalkyl substances (PFAS)?

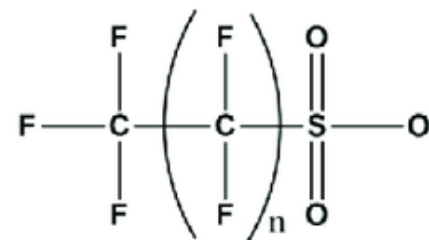
3

PFAS are a group of 12,000+¹ man-made chemicals containing carbon-fluorine bonds that do not degrade easily in the environment or in living tissues, earning them the nickname “**forever chemicals**.”

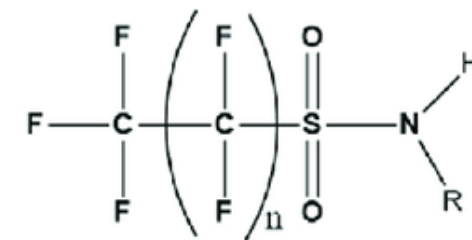
C-F bond = 466 kJ mol⁻¹



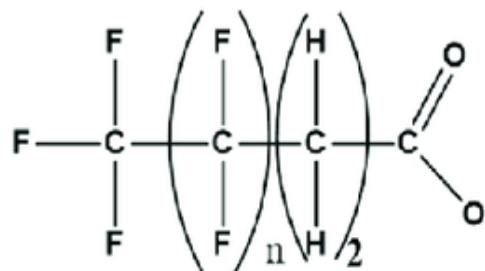
(a) Perfluoroalkyl carboxylate



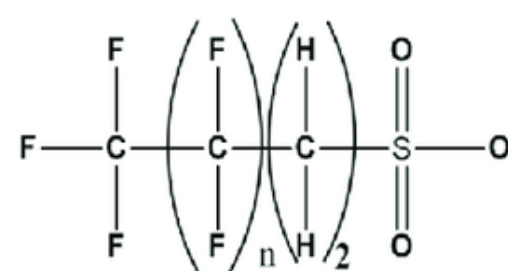
(b) perfluoroalkyl sulfonate



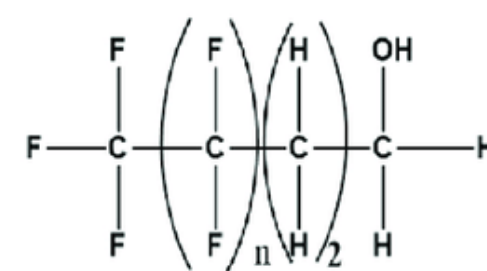
(c) perfluoroalkyl sulfonamide



(d) Polyfluoroalkyl carboxylate



(e) Polyfluoroalkyl sulfonates



(f) Polyfluoroalkyl carboxylate

Olatunde et al. (2020), *Heliyon*

¹USEPA (2021). *PFAS Master List of PFAS Substances (RETIRED)*. CompTox chemicals dashboard. <https://comptox.epa.gov/dashboard/chemical-lists/PFASMASTER>

PFAS are found in products we come in contact with everyday.



PFAS: Last Week Tonight with John Oliver (HBO)

YouTube | LastWeekTonight | 5.7M views | Oct 4, 2021

[PFAS: Last Week Tonight with John Oliver \(HBO\) - YouTube](#)



John Wiegand, MIT Technology Review



Progressive Charlestown

Toxic Carpet: We're Breathing Harmful Forever Chemicals in Homes, Offices, and Classrooms

TOPICS: Pollution Popular Public Health
By GREEN SCIENCE POLICY INSTITUTE AUGUST 31, 2021



SciTechDaily.com

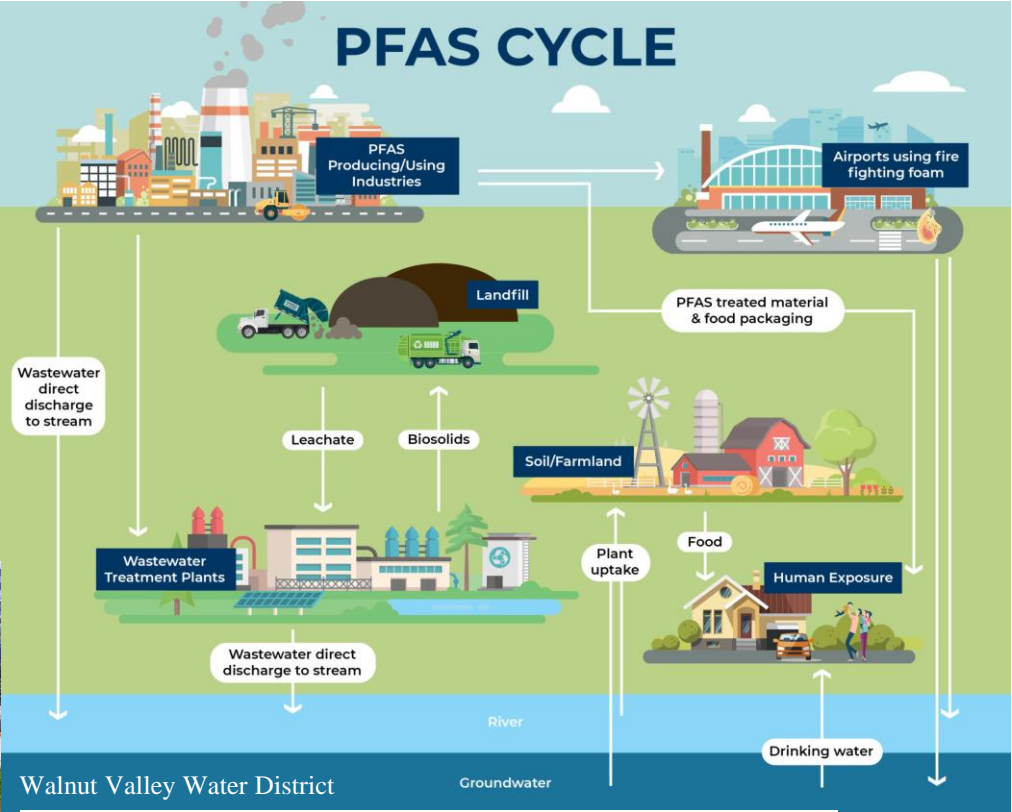
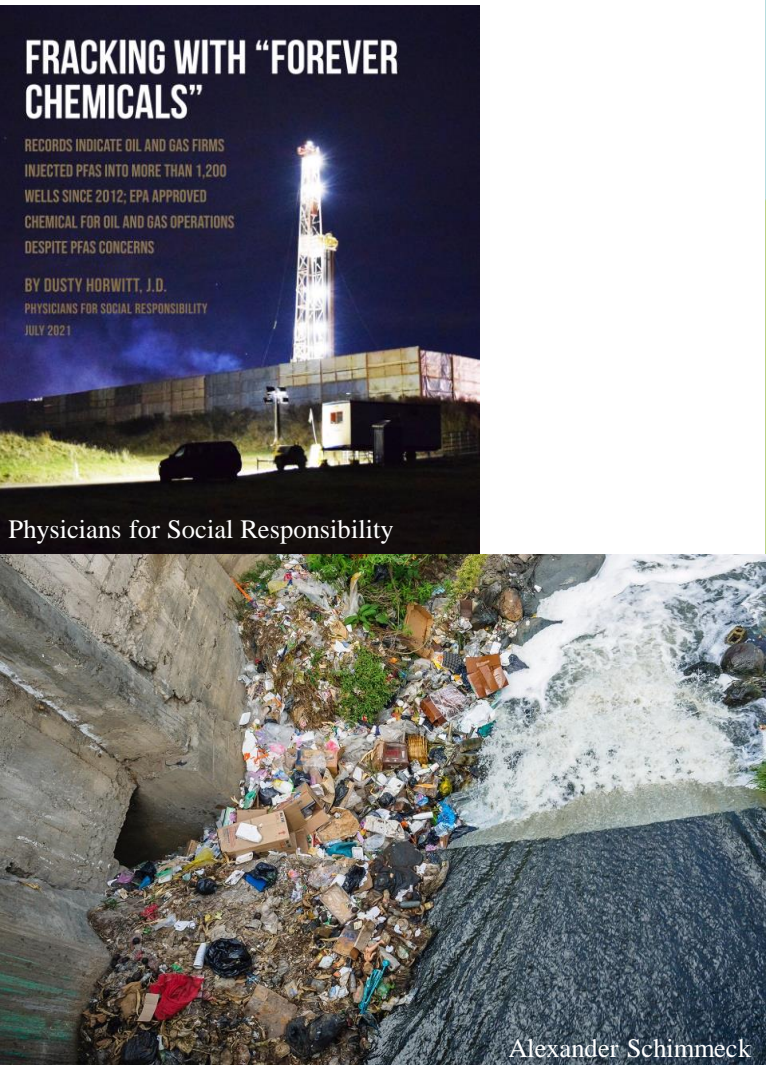
NEWS & INSIGHTS / NEWS / 2022 / 01

New tests find toxic 'forever chemicals' in bedding, yoga pants and other textiles



ewg.org

PFAS can also be found in drinking water, fruits, vegetables, meat, lakes, rivers, and soil due to their resistance to degradation and tendency to accumulate in the environment.



PUBLIC INTEREST

PFAS poisoning: Don't eat deer from near Clark's Marsh

Published: Nov. 06, 2023, 1:30 p.m.

MLive

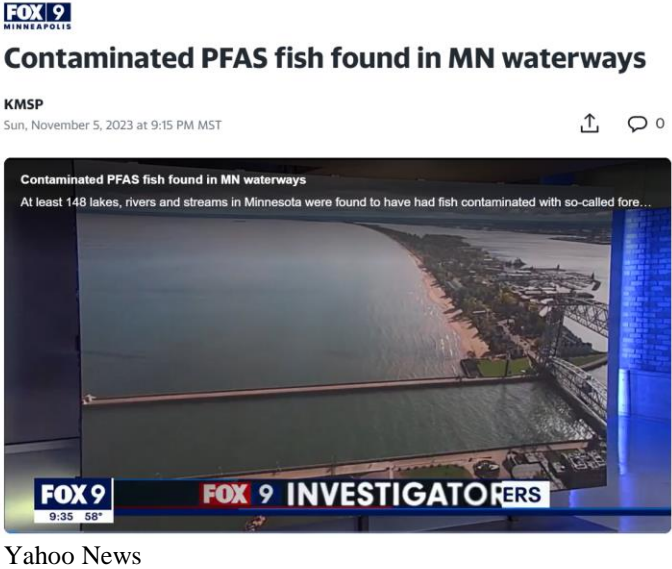
PFAS 'forever chemicals' found in 71% of Wisconsin's shallow, private wells

450 shallow, private wells were sampled in study

Associated Press

Published November 3, 2023 5:28pm EDT

Fox News

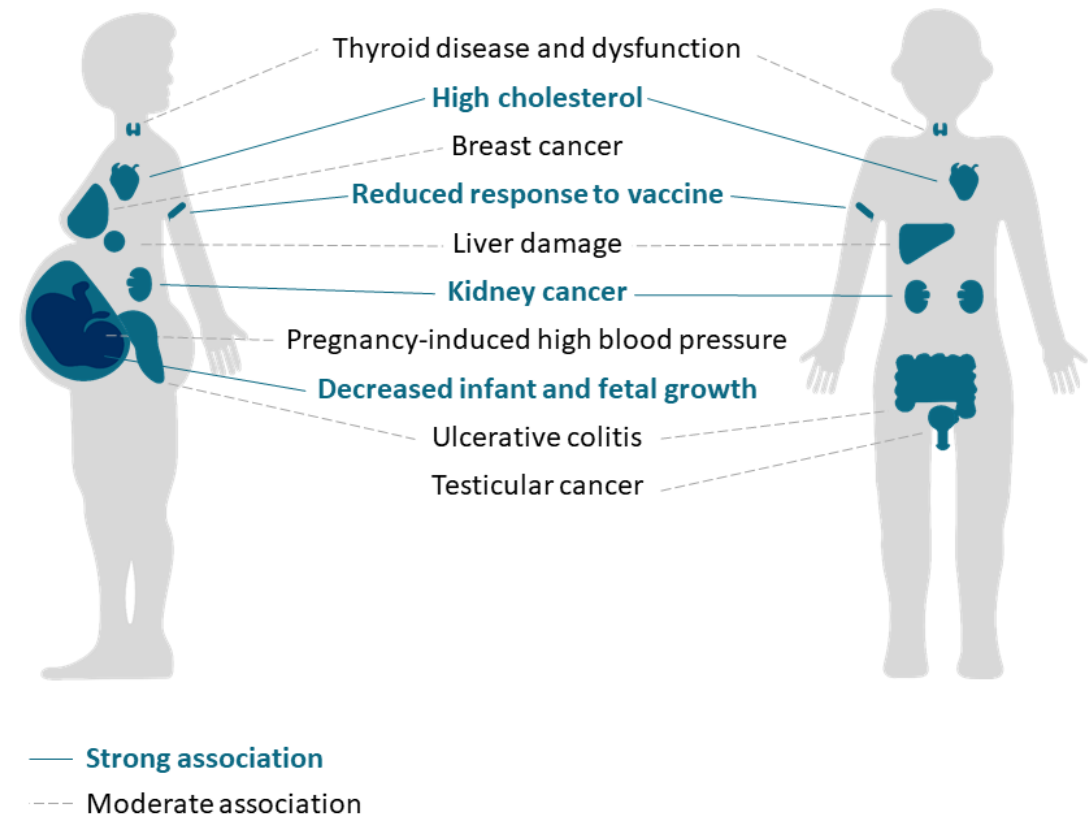


Motivation to quantify and monitor PFAS

Exposure to certain levels of PFAS may result in the following¹:

- Decreased fertility
- Increased high blood pressure for pregnant women
- Developmental abnormalities in children
 - low birth weight
 - accelerated puberty
 - bone variations,
 - behavioral changes
- Increased risk of prostate, kidney, and testicular cancer
- Reduced immune system functioning
- Reduced response to vaccines
- Interference with the body's natural hormones
- Increased cholesterol and/or obesity.

Health Effects of PFAS



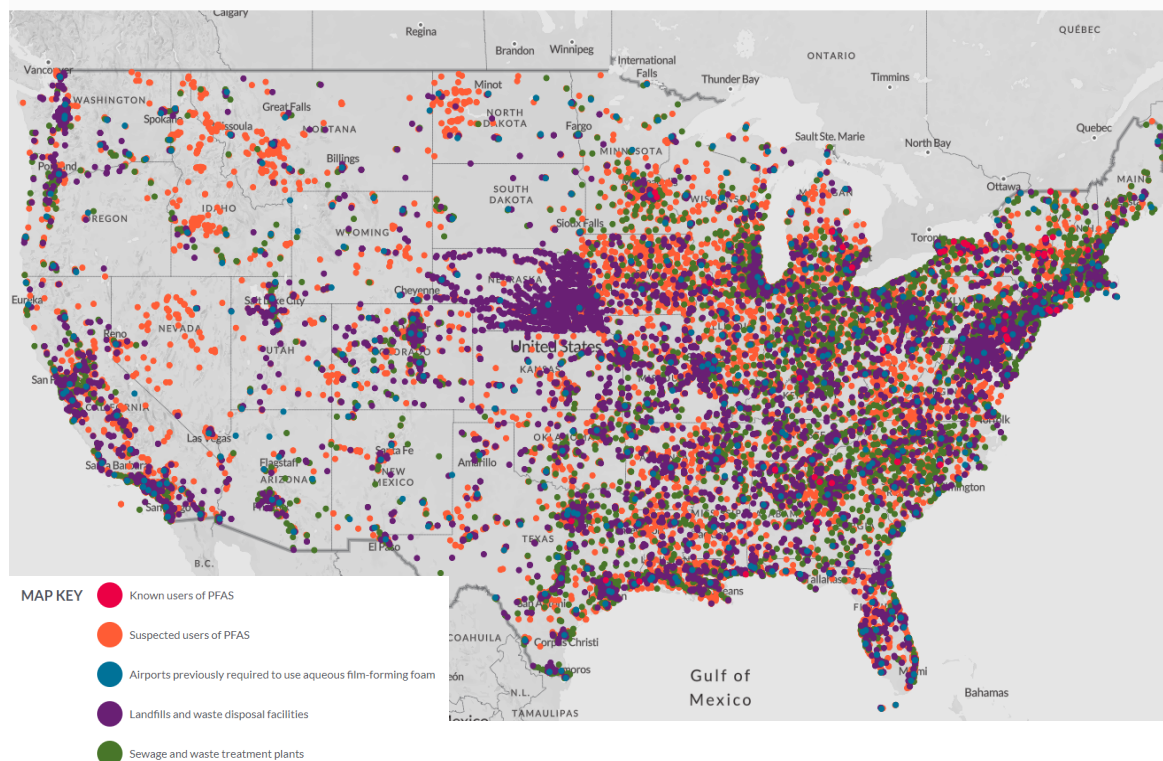
Madison and Dane County Public Health

¹USEPA (2023, June 03). *Our Current Understanding of the Human Health and Environmental Risks of PFAS*. United States Environmental Protection Agency. <https://www.epa.gov/pfas/our-current-understanding-human-health-and-environmental-risks-pfas>

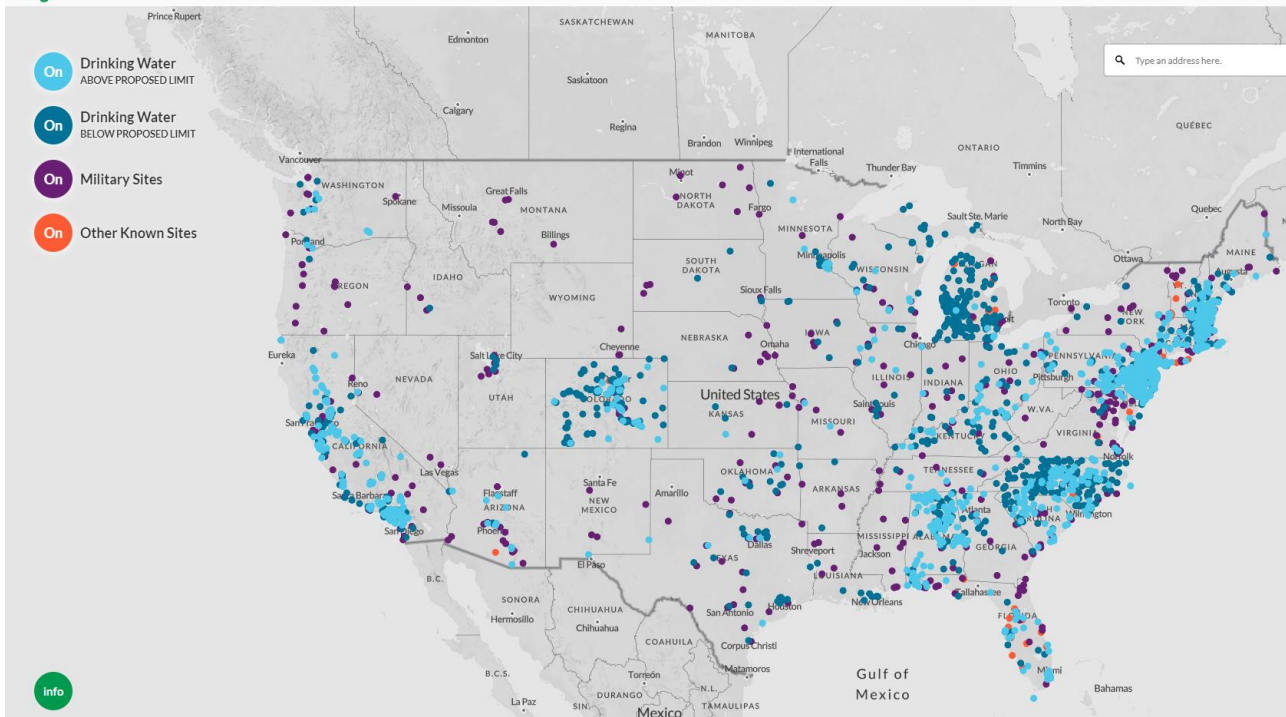


The United States East Coast has numerous suspected PFAS sources and instances of drinking water contamination.

Suspected industrial discharges of PFAS



PFAS Contamination in the U.S. (August 17, 2023)



Airborne PFAS can come from three main sources.

1. Direct emission from point sources

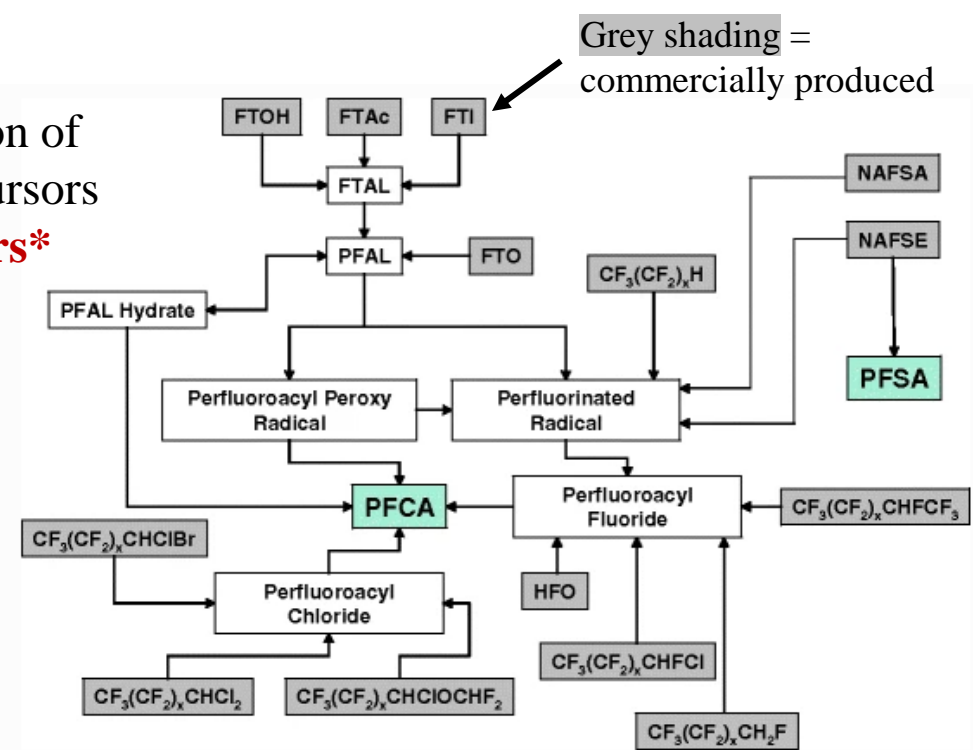
~Days – weeks*



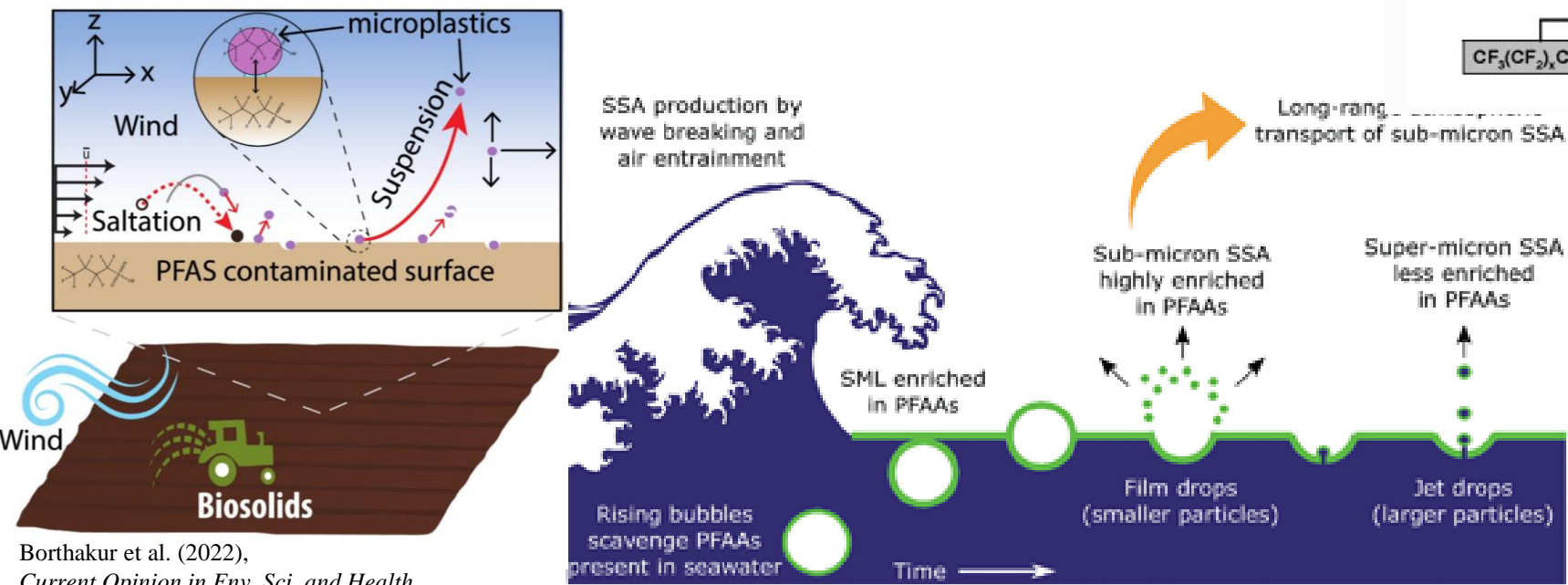
toxicfreefuture.org

3. Degradation of volatile precursors

~Days – years*



Young and Mabury (2010), *Reviews of Environmental Contamination and Toxicology*



Borthakur et al. (2022), *Current Opinion in Env. Sci. and Health*

2. Dust and sea spray aerosols emitted from soil and bodies of water contaminated with PFAS.

~Days – weeks*

***Atmospheric lifetime of relevant PFAS**

1. Atmospheric aerosol particles

Environmental
Science
Processes & Impacts

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PFAS on atmospheric aerosol particles: a review

Cite this: *Environ. Sci.: Processes Impacts*, 2023, 25, 133

Jennifer A. Faust 


Table 4 PFOA and PFOS in atmospheric particulate matter (pg m ⁻³) ^a						
Ref.	Year ^b	Site	Type ^c	Size ^d (µm)	PFOA	PFOS
110	2001–2002	Fukuchiyama, Japan	U	—	—	0.6 ± 1.3
110	2001–2002	Oyama, Japan	U	—	—	5.3 ± 1.2
111	2001–2002	Oyama, Japan	U	—	262.7 ± 2.4	5.2
111	2003	Morioka, Japan	U	—	2.0 ± 1.2	0.7
7	2003–2004	Parkersburg, WV, USA	I	<0.28 to >4.0	<3000–1 × 10 ⁶	—
15	2005	Fukuchiyama, Japan	U	<0.46 to 100	15.2	2.2
15	2005	Oyama, Japan	U	<0.46 to 100	205	2.9
15	2005	Route 171, Japan	U	<0.46 to 100	330	6.8
112	2005	Lakes Erie & Ontario, N. America	RM/SU	—	—	nd ^e –8.1
70	2005	Atlantic Ocean: Germany to S. Africa	M	—	0.3–1.5	0.05–2.5
96	2005	Kjeller, Norway	RR	—	1.43–1.67	0.89–1.13
96	2005	Manchester, UK	U	—	15.7–455	7.1–51
96	2005–2006	Hasleberg, UK	RR	—	6.9–828	0.9–9.28
96	2006	Mace Head, Ireland	C/RR	—	3.1–16	<1.8
97	2006	Albany, NY, USA	U	—	0.76–4.19	0.35–1.16
99	2007	North Sea, Germany	M	—	1.9–6.1	0.1–2.3
101	2007–2008	Bardsholte, Germany	SR	—	0.3 ± 0.4	1.3 ± 2.5
101	2007–2008	Geesthacht, Germany	SR	—	0.2 ± 0.4	0.6 ± 1.9
66	2007–2008	Geesthacht, Germany	SR	<0.14 to 11.4	0.1–4.8	0.2–3.5
102	2009	Lüneburg and Löhne, Germany	SR/SU	—	nd–3.4	nd–2.9
10	2009	Northern Germany	U/RR/SU	—	nd–1.8	nd–1.3
103	2010	Alcánte, Spain	U/RR/SU	<2.5	1.4–13.8	<1.4 to 4.4
104	2010	Zürich, Switzerland	U	<70	7.7	2.3
104	2010	Mt. Uetliberg, Switzerland	RM	<70	1.7	1.7
50	2010	Toronto, Canada	SU	1–25	nd–0.47	nd–4.8
107	2012–2013	Karachi, Pakistan	C/U/U	—	0.85–8.70	0.64–3.17
52	2012–2014	Kiel, Czech Republic	RR	<10	nd–0.68	nd–0.59
58	2013	Beijing, China	CU	<2.5	0–10.4	0.2–8.1
64	2013–2015	Shanghai, China	C/U	<0.4 to >9.0	219	48.6
31	2014 ^f	Weifang, China	I	>0.7	3.79–3820	3.70–68.1
31	2014 ^g	Weifang, China	I	>0.7	19.0–1530	2.12–154
31	2014 ^h	Tianjin, China	C/I	>0.7	17.3–90.5	4.69–3830
31	2014 ⁱ	Tianjin, China	C/I	>0.7	1.30–360	1.57–19.3
31	2014 ^j	N. Huangcheng Island, China	M	>0.7	3.53–223	3.53–182
31	2014 ^k	N. Huangcheng Island, China	M	>0.7	9.65–233	2.30–41.2
21	2014–2015	Beijing, China	U	<2.5 and >2.5	12.5	0.8
21	2014–2015	Changshu, China	U	<2.5 and >2.5	2.07	0.14
21	2014–2015	Guyang, China	U	<2.5 and >2.5	325	6.28
21	2014–2015	Ji'nan, China	U	<2.5 and >2.5	335	6.28
21	2014–2015	Nanjing, China	U	<2.5 and >2.5	11.6	4.3
71	2016	Bohai and Yellow Seas, China	M	>0.7	3.5–43	1.1–6.4
16	2014–2016	Tsukuba, Japan	SU	<0.5 to >10	1.3–5.4	nd–0.6
33	2018	S. Shetland Islands, Antarctica	M/RR	0.1–2	nd–0.04	nd–0.025
67	2018–2019	Xiamen, China	C/U	<1 to >10	nd–12.23	nd–45.66
72	2018–2019	Birkenes, Norway	M	—	<0.003–0.811	0.006–0.392
72	2018–2020	Andøya, Norway	M	—	<0.003–1.28	<0.004–0.144
88	2019	N. Carolina, USA	RS	<2.5	nq–14.06	nq–4.75
60	2019–2020	Dongshan Island, China	M	<1 to >10	2.6–4.5	5.0–8.1
108	2020	Tsukuba, Japan	SU	>1 to >10	—	0.55

^a Values are reported as a range when minimum and maximum concentrations were available. Single values are mean concentrations with standard deviation, if available. ^b Year refers to time of measurements, not publications. ^c C – coastal, I – industrial, M – marine, RM – remote, RR – rural, RS – residential, SR – semi-rural, SU – suburban, U – urban. ^d Diameter ranges of particles sampled. ^e nd – less than limit of detection. ^f nq – less than limit of quantitation. ^g s – summer. ^h w – winter.

Summarizes results from 37 studies from 2001 - 2020.

2. Wet deposition

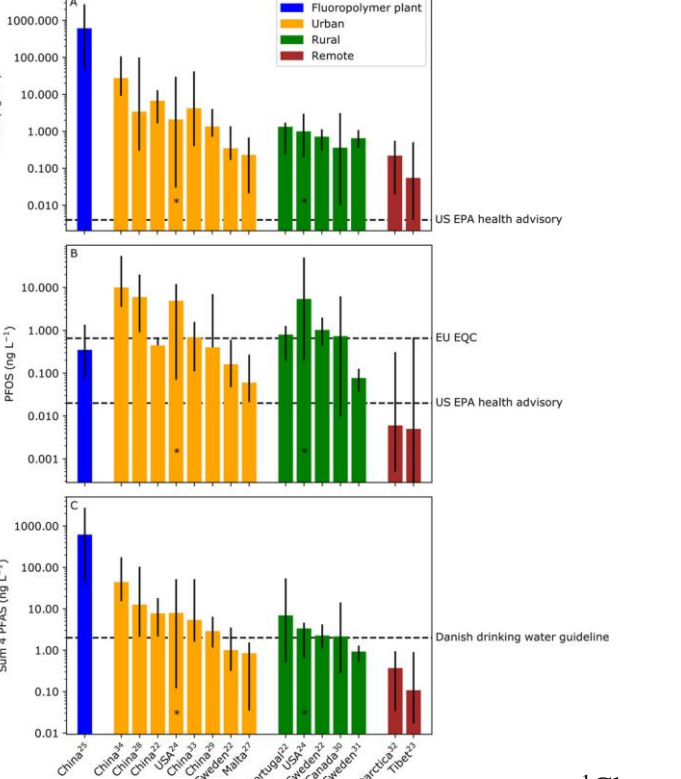
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Outside the Safe Operating Space of a New Planetary Boundary for Per- and Polyfluoroalkyl Substances (PFAS)

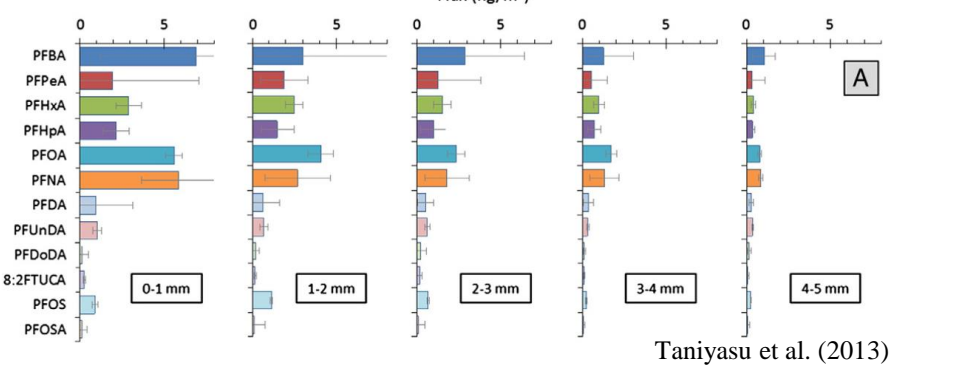
Ian T. Cousins,* Jana H. Johansson, Matthew E. Salter, Bo Sha, and Martin Scheringer



Summarizes results from 11 studies from 2010 - 2022.

3. Cloud droplets? Here's what we know:

- Shorter chain PFAS are water soluble.¹
- PFAS reduce surface tension² yet lower vapor pressure of water.³
- Lab studies show PFOA have ice nucleating ability.⁴
- PFAS concentrations in rainwater can vary with duration of the rain event.^{5,6}
- Modeling studies have attempted to simulate nucleation and scavenging of PFAS-containing particles in clouds.^{7,8}



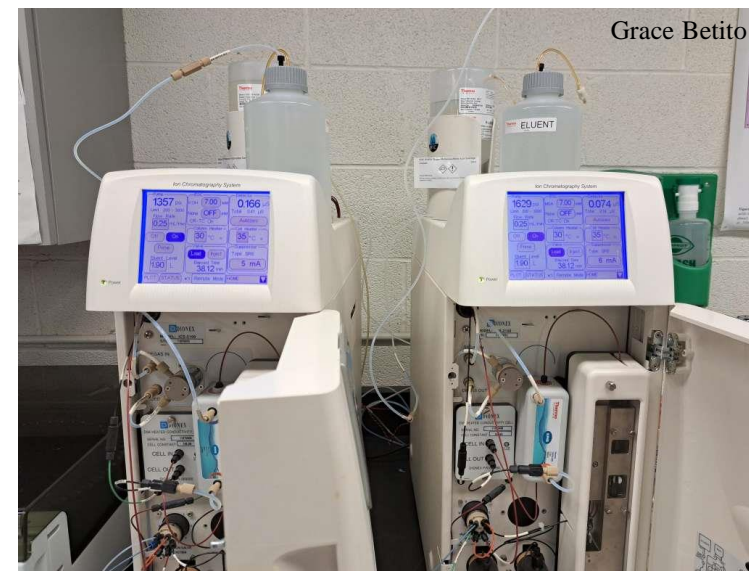
Taniyasu et al. (2013)

¹Chen et al. (2019), *Water Research*, ²Buck et al. (2012), Faust et al. (2023), *Env. Sci.: Processes and Impacts*, ⁴Schwidetzky et al. (2021), *J. Phys. Chem. Letters*, ⁵Kwok et al. (2010), *Env. Sci. & Technology*, ⁶Taniyasu et al. (2013), *Env. International*, ⁷D'Ambro et al. (2023), *Sci. of the Total Env.*, ⁸Franco et al. (2011), *Chemosphere*



Cloud water samples from ACTIVATE 2020 and 2021 flights were analyzed for mass concentrations of water-soluble ions, elements, and PFAS.

1. Cloud water collected using the AC3.



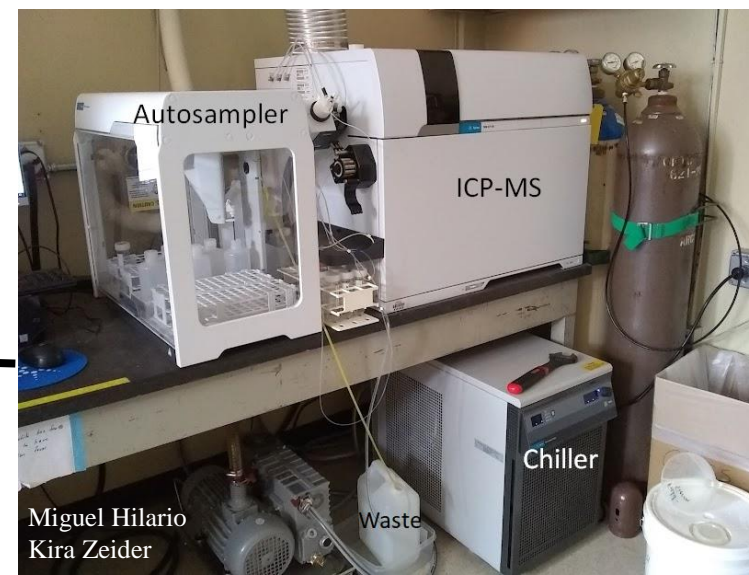
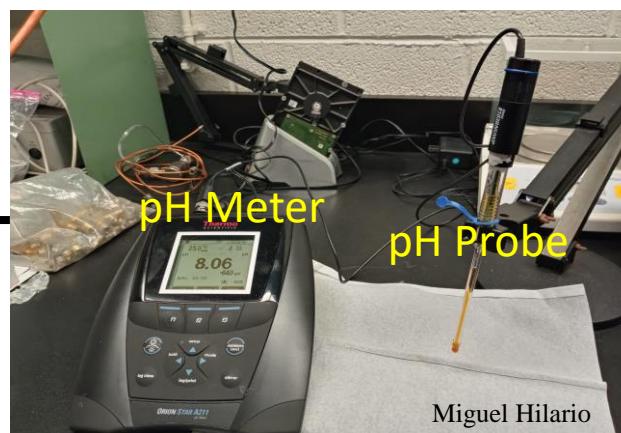
2. IC system provided aqueous water-soluble ion mass concentrations.

IC = ion chromatography

5. Agilent 6470 QQQ LC-MS/MS provided aqueous PFAS mass concentrations.



4. pH analysis

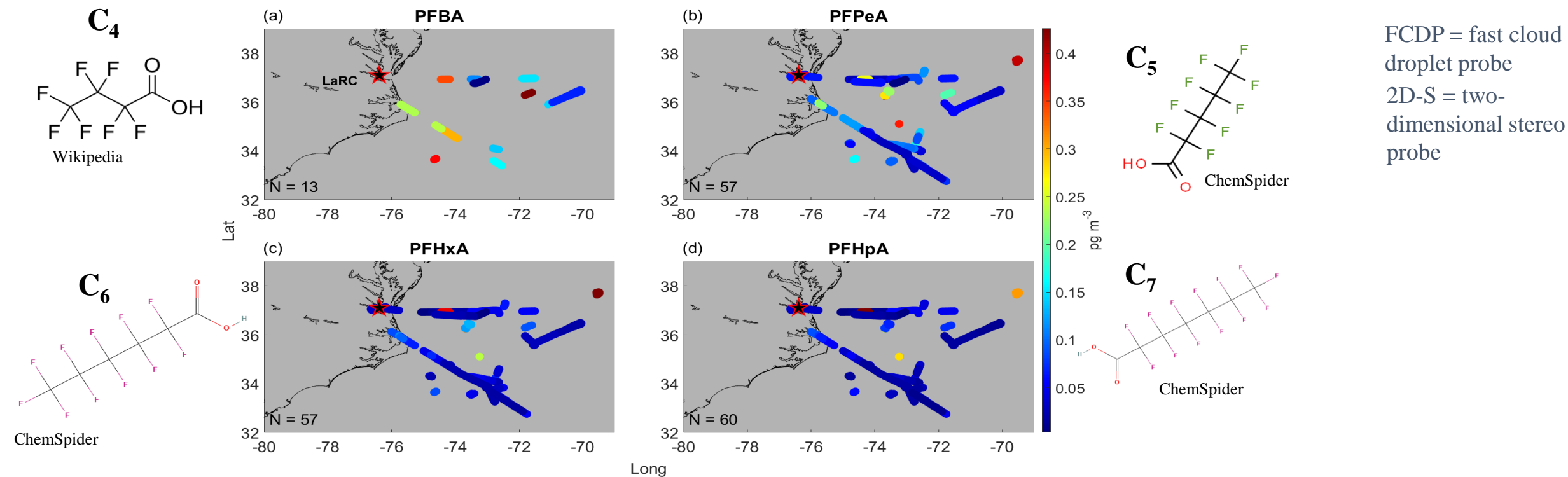


3. ICP-MS system provided aqueous water-soluble elemental mass concentrations.

ICP-MS = inductively coupled plasma mass spectrometry



Aqueous mass concentrations of ions, elements, and PFAS were converted to air-equivalent using the mean total water content (TWC; FCDP + 2D-S) for periods when $\text{TWC} > 0.02 \text{ g m}^{-3}$ for each sample.



- PFBA, PFPeA, PFHxA, and PFHpA detected in 13, 57, 57, and 60 samples (out of 62), respectively.
- PFBA mass concentrations typically higher than C₅ – C₇ PFAS, consistent with past works.^{1,2}
- Mass concentrations for C₄ – C₇ PFAS are an order of magnitude lower than those reported **in aerosol particles** over China, India, Japan, and South Korea³ as well as Toronto, Canada¹ and Northwest Europe⁴.
- Values for PFHpA are an order of magnitude higher than those reported **in PM_{2.5}** over North Carolina, U.S.⁵



Sources/particle types associated with PFAS

		62	62	62	62	59	59	58	62	60	23	62	60	62	59	59	41	13	57	57	60
ssNa ⁺																					
ssCa ²⁺	0.99***		62	62	62	59	59	58	62	60	23	62	60	62	59	59	41	13	57	57	60
Cl ⁻	0.98***	0.96***		62	62	59	59	58	62	60	23	62	60	62	59	59	41	13	57	57	60
dust Na ⁺	-0.19	-0.07	-0.20		62	59	59	58	62	60	23	62	60	62	59	59	41	13	57	57	60
dust Ca ²⁺	-0.19	-0.07	-0.20	1.00***		59	59	58	62	60	23	62	60	62	59	59	41	13	57	57	60
Al	-0.29*	0.37**	0.25	0.51***	0.51***		59	58	59	57	23	59	57	59	59	59	41	13	54	54	57
Mn	-0.36**	0.48***	0.34**	0.65***	0.65***	0.68***		58	59	57	23	59	57	59	59	59	41	13	54	54	57
Fe	-0.38**	0.46***	0.38**	0.30*	0.30*	0.77***	0.46***		58	56	23	58	56	58	58	58	41	13	53	53	56
NO ₃ ⁻	-0.45***	0.52***	0.46***	0.47***	0.47***	0.86***	0.78***	0.66***		60	23	62	60	62	59	59	41	13	57	57	60
Oxalate	-0.09	0.12	0.08	0.33**	0.34**	0.68***	0.34*	0.43***	0.54***		22	60	58	60	57	57	39	12	55	55	58
MSA	-0.43*	0.49*	0.40	0.32	0.32	0.49*	0.64**	0.13	0.60**	0.25		23	22	23	23	23	16	4	21	22	23
NH ₄ ⁺	-0.34**	0.38**	0.34**	0.32*	0.32*	0.63***	0.59***	0.56***	0.76***	0.38**	0.36		60	62	59	59	41	13	57	57	60
K ⁺	0.95***	0.93***	0.95***	-0.16	-0.16	0.28*	0.40**	0.33*	0.53***	0.10	0.59**	0.42**		60	57	57	41	13	55	55	58
nssSO ₄ ²⁻	-0.59***	0.63***	0.62***	0.22	0.22	0.57***	0.63***	0.44***	0.81***	0.37**	0.66***	0.68***	0.70***		59	59	41	13	57	57	60
V	-0.22	0.31*	0.20	0.55***	0.55***	0.76***	0.59***	0.63***	0.68***	0.56***	0.43*	0.51***	0.18	0.51***		59	41	13	54	54	57
Cr	-0.24	0.28*	0.18	0.35**	0.35**	0.71***	0.52***	0.34**	0.72***	0.39**	0.52*	0.48***	0.30*	0.42***	0.38**		41	13	54	54	57
As	-0.07	0.20	0.07	0.60***	0.60***	0.74***	0.61***	0.65***	0.59***	0.48**	0.30	0.53***	0.01	0.37*	0.85***	0.28		12	37	39	40
PFBA	-0.07	0.14	0.07	0.49	0.52	0.77**	0.62*	0.69*	0.75**	0.65*	-0.80	0.65*	-0.01	0.52	0.85***	0.78**	0.86***		13	13	13
PFPeA	-0.02	0.11	-0.07	0.66***	0.66***	0.60***	0.59***	0.39**	0.46***	0.46***	0.27	0.36**	-0.01	0.20	0.57***	0.39**	0.74***	0.96***		53	55
PFHxA	-0.06	0.18	-0.03	0.70***	0.70***	0.71***	0.74***	0.43**	0.65***	0.43***	0.55**	0.46***	0.09	0.34**	0.57***	0.60***	0.69***	0.91***	0.95***		57
PFHpA	-0.10	0.21	0.02	0.71***	0.71***	0.69***	0.74***	0.38**	0.64***	0.38**	0.63**	0.42***	0.14	0.36**	0.56***	0.61***	0.63***	0.79**	0.86***	0.96***	
	ssNa ⁺	ssCa ²⁺	Cl ⁻	dust Na ⁺	dust Ca ²⁺	Al	Mn	Fe	NO ₃ ⁻	Oxalate	MSA	NH ₄ ⁺	K ⁺	nssSO ₄ ²⁻	V	Cr	As	PFBA	PFPeA	PFHxA	PFHpA

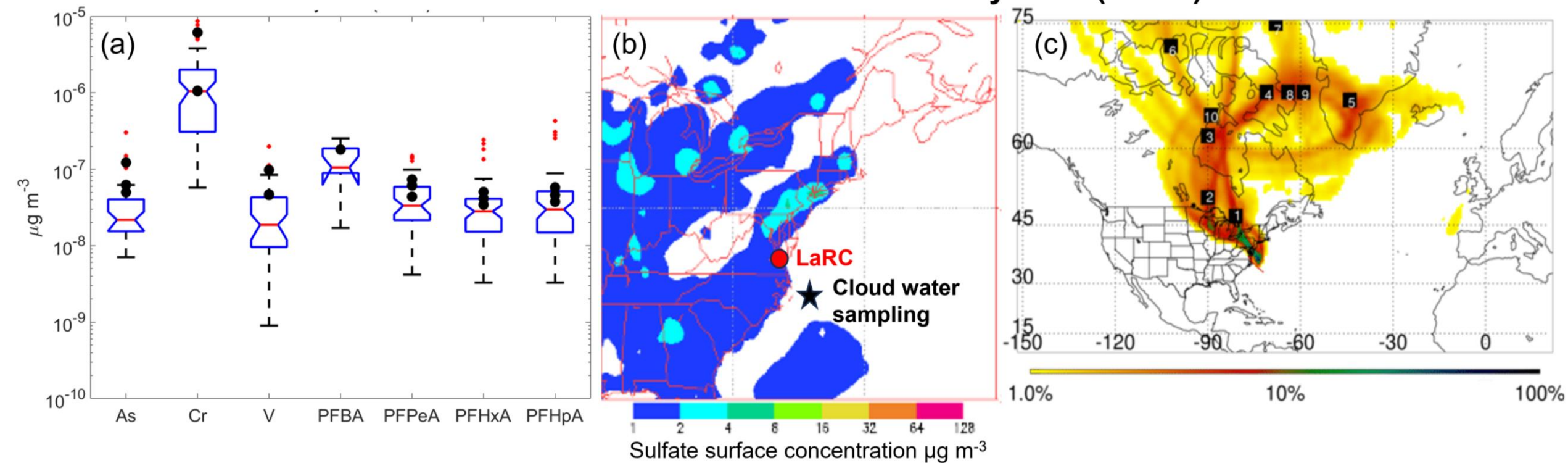
- C₄ – C₇ PFAS correlated with As, V, and Cr, suggesting combustion influence.¹
- C₅ – C₇ PFAS correlated with dust Na⁺, dust Ca²⁺, Al, Mn, NO₃⁻, oxalate, and MSA, indicating association with dust particles.^{1,2,3,4}
- Weak correlations between C₄ – C₇ PFAS and ssNa⁺, ssCa²⁺, and Cl⁻ despite past works identifying relationships between PFAS and sea salt.^{5,6}

Colored boxes show Spearman rho values, white boxes show number of samples considered for each correlation.

* = p < 0.05, ** = p < 0.01, *** = p < 0.001



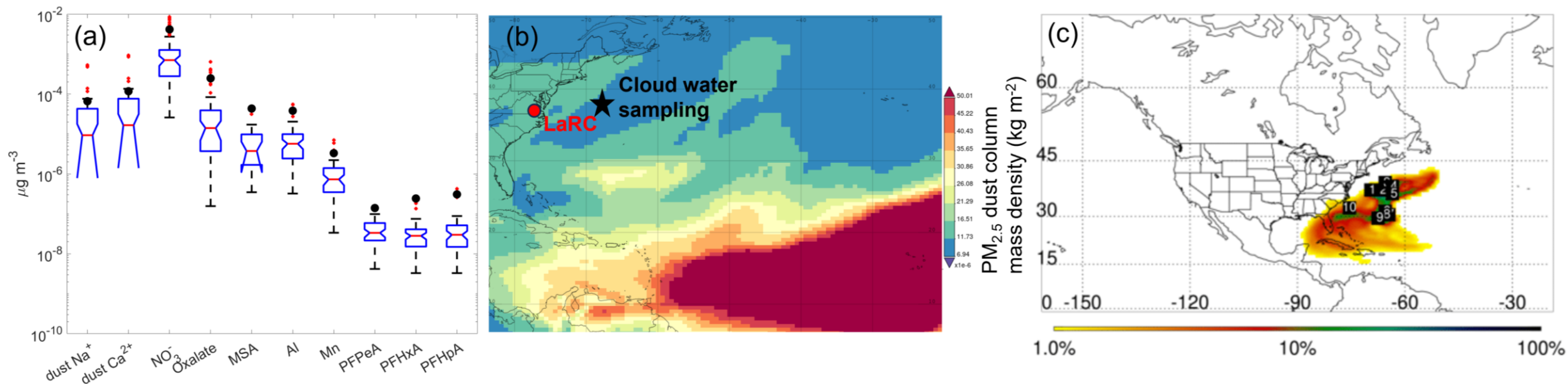
Combustion influence – 15 February 2020 (RF 02)



- As, Cr, V, and $\text{C}_4 - \text{C}_7$ PFAS mass concentrations are mostly above their respective medians.
- NAAPS simulates sulfate as the only particle type contributing to AOD over the Upper Midwest and Northeast where FLEXPART back trajectories come from.
- $\text{C}_4 - \text{C}_7$ PFAS in cloud water appear to be associated with combustion particles originating from these regions, consistent with past works.^{1,2}
- PFAS can readily adsorb to organic matter and soot particles when they are protonated (i.e., at low pH).¹



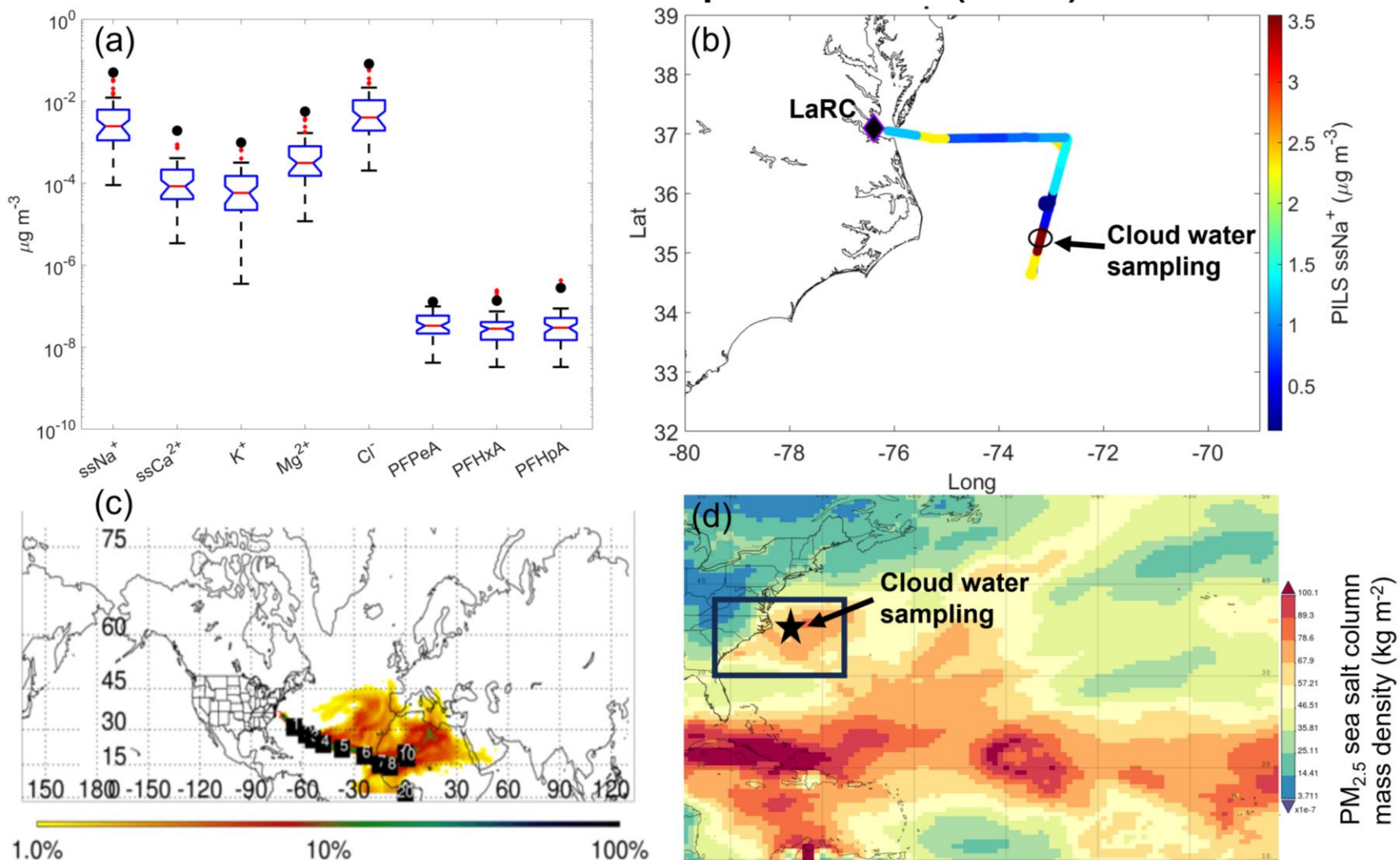
Dust influence – 28 June 2021 (RF 90)



- High mass concentrations observed for dust tracers and $\text{C}_5 - \text{C}_7$ PFAS.
- MERRA-2 shows African dust over Caribbean and Atlantic from June 19 – 28, 2021 where FLEXPART back trajectories come from.
- PFAS appear to be associated with African dust particles.
- There are currently no studies discussing PFAS in African dust plumes, yet past works have found PFAS to adsorb to the surface of dust particles and be transported long distances.^{1,2}



Sea salt influence – 10 September 2020 (RF 32)



- High mass concentrations observed for sea salt tracers and $\text{C}_5 - \text{C}_7$ PFAS.
- PILS Na^+ also high where the sample was obtained.
- FLEXPART back trajectories show sampled air mass originated over the Atlantic where MERRA-2 $\text{PM}_{2.5}$ sea salt column density values are high.
- Many past works have found sea spray aerosols to contain and transport PFAS over long distances.^{1,2}

PILS = particle into liquid sampler



- This is the first study quantifying PFAS mass concentrations in cloud water.
- PFBA, PFPeA, PFHxA, and PFHpA are present at detectable levels.
- PFAS are most correlated with tracers for combustion and dust.
- Case studies show elevated PFAS associated with the following:
 - Combustion particles from the Upper Midwest and Northeastern U.S.
 - African dust
 - Marine air with relatively high sea salt mass concentrations
- These results have the following implications:
 - Demonstrate cloud droplets play a role in how PFAS are distributed globally.
 - Show PFAS in rainwater can come from both cloud droplets and below-cloud scavenging.
 - Provide validation data for past and future modeling studies simulating airborne PFAS life cycles and their influence on cloud droplet thermodynamics.

Questions?
evalouedwards@arizona.edu

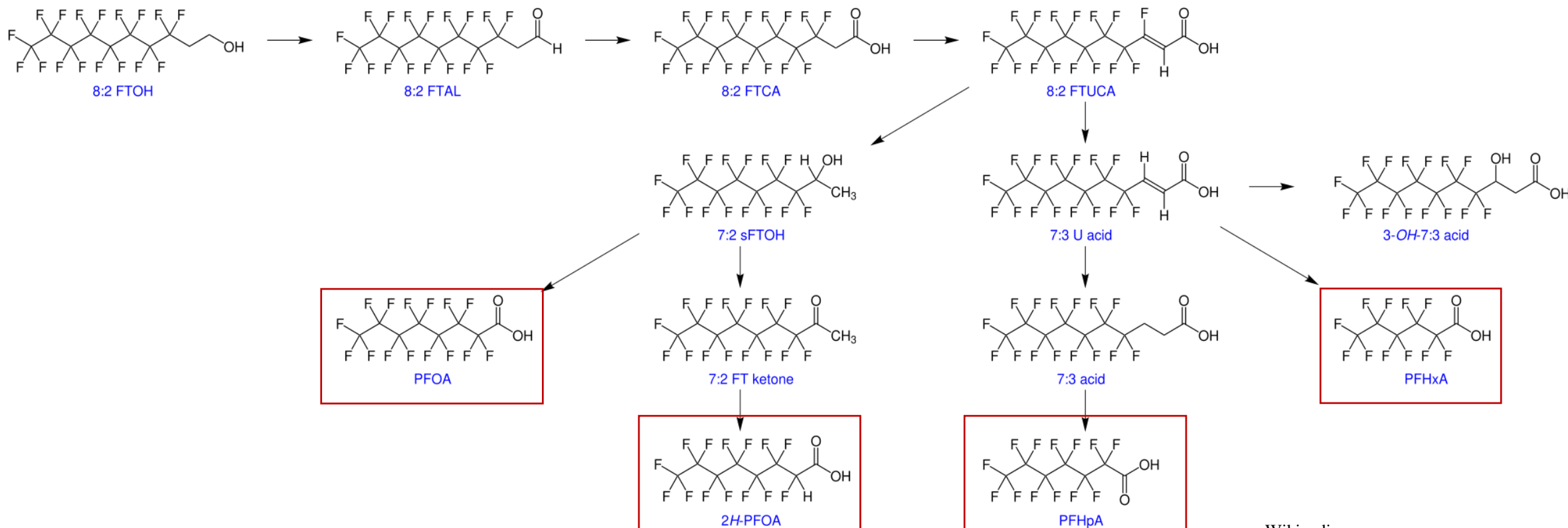


Backup Slides

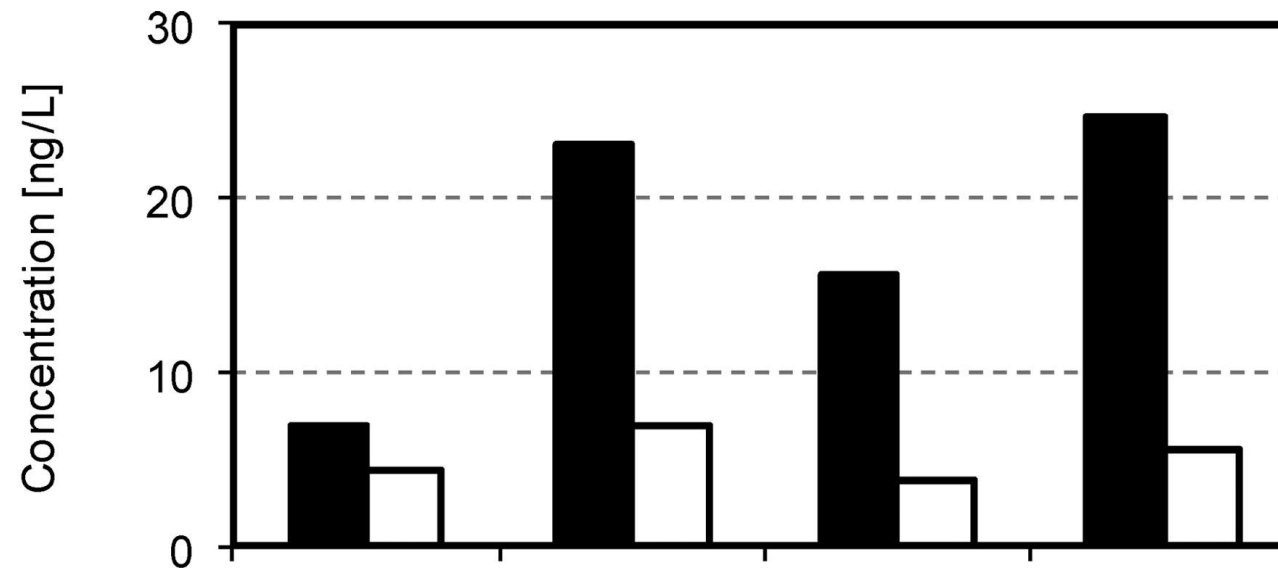


2. Degradation of volatile precursors (e.g., fluorotelomer alcohols [FTOHs])

At least 20 days

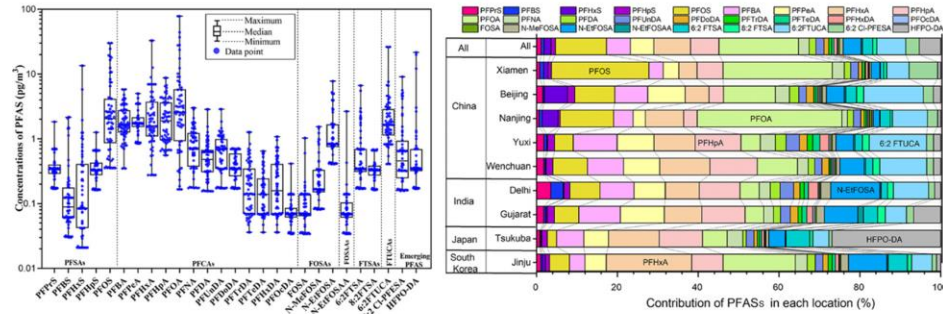


Wikipedia

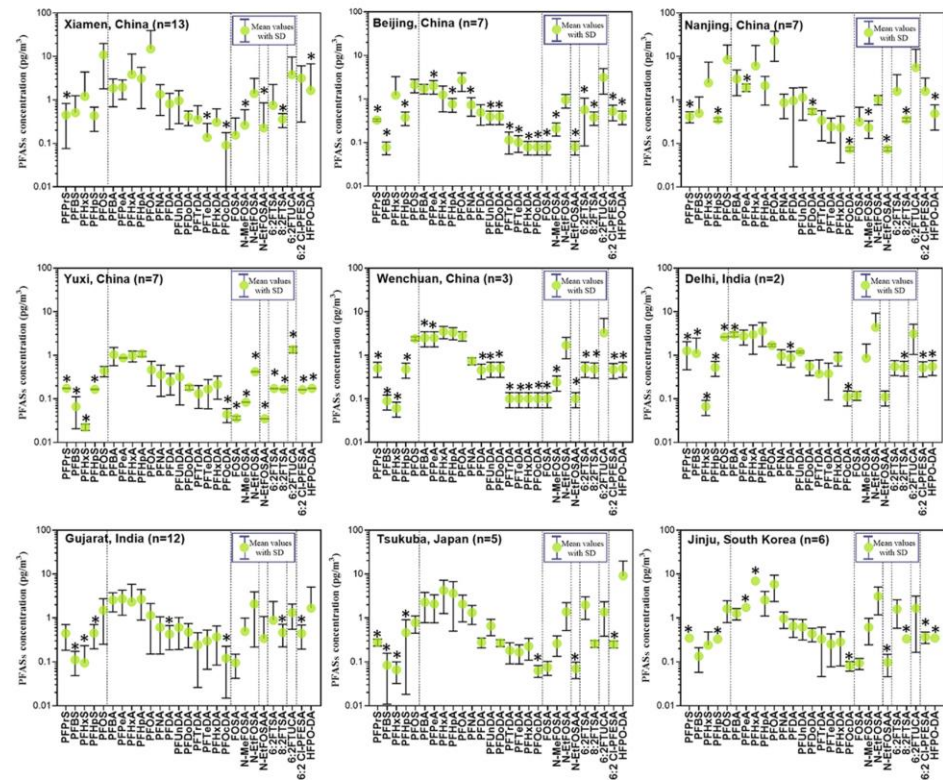


Day	23 rd 24 th	29 th 30 th 30 th 31 st	19 th 19 th 20 th	11 th 12 th 12 th 13 th
Month Year	Oct 2006	Jul 2007	Oct 2007	Jan 2008
Location	Tsukuba	Tsukuba	Kawaguchi	Kawaguchi
Precipitation [mm]	22.4 9.6	37.4 20.1	7.4 9.6	5.7 4.5

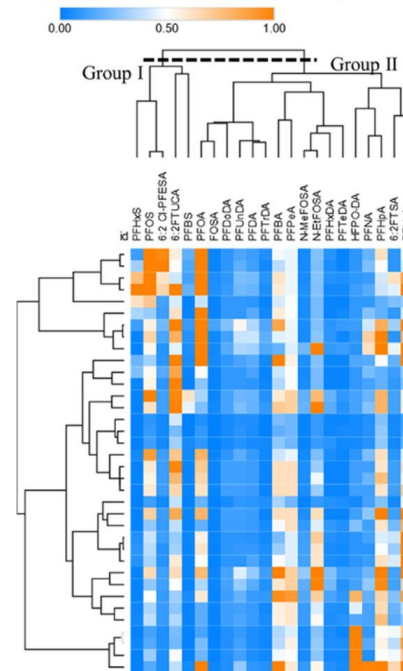
(b) Composition profiles



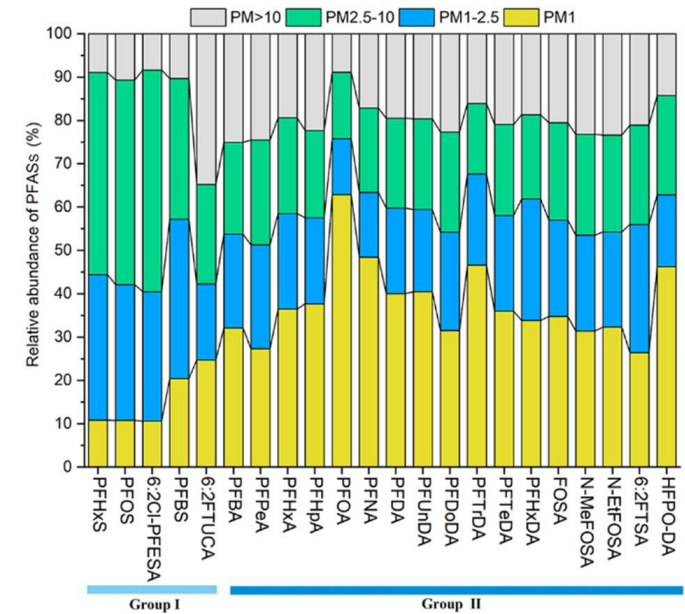
(c) Mean concentrations and standard deviations in each location



(a) Hierarchical cluster analysis



(b) Relative abundance of PFASs on different size fractions



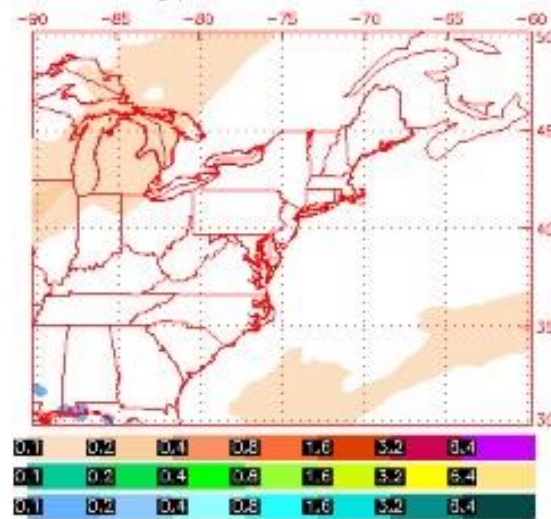


PFAS in ACTIVATE cloud water

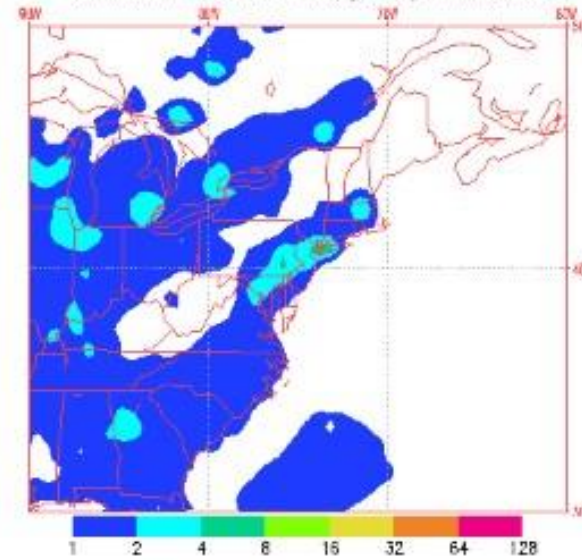
Statistic	Units	PFBA	PFPeA	PFHxA	PFHpA
N samples		13	57	57	60
Median	pg m ⁻³	0.11	0.03	0.03	0.03
Mean		0.13	0.04	0.04	0.05
Std. Deviation		0.07	0.03	0.05	0.08
Minimum		0.02	0.004	0.003	0.003
Maximum		0.25	0.15	0.24	0.43



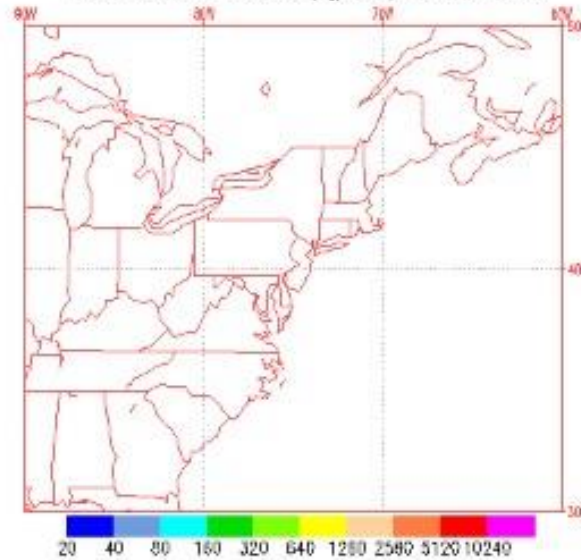
NAAPS Total Optical Depth for 18:00Z 15 Feb 2020
Sulfate: Orange/Red, Dust: Green/Yellow, Smoke: Blue



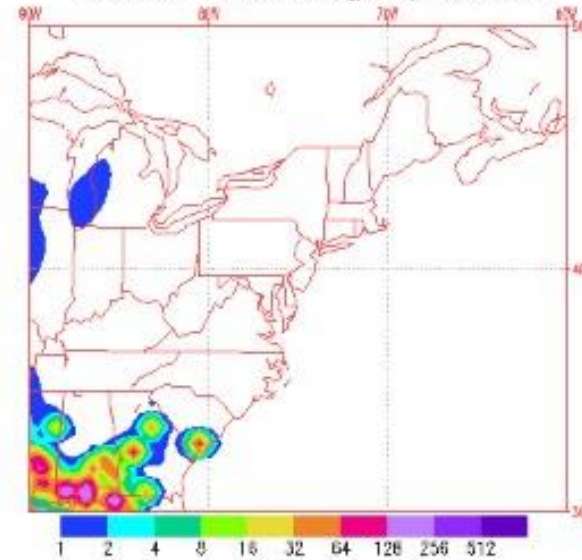
Sulfate Surface Concentration ($\mu\text{g}/\text{m}^3$) for 2020021518



Dust Surface Concentration ($\mu\text{g}/\text{m}^3$) for 2020021518



Smoke Surface Concentration ($\mu\text{g}/\text{m}^3$) for 2020021518





- What are per- and polyfluoroalkyl substances (PFAS)?
- Sources of PFAS
- Motivation to quantify and monitor PFAS in the environment
- Current understanding of airborne PFAS
- Methods
- PFAS in ACTIVATE cloud water samples
- Sources/particle types associated with PFAS
- Case studies
- Conclusions and discussion